

Mineral	Composition	Specific gravity	Magnetic response			Electrostatic response	
			F	P	NM	C	NC
Actinolite	$\text{Ca}_2(\text{Mg}, \text{Fe})_3[\text{Si}_4\text{O}_{11}]_2(\text{OH})_2$	3.0-3.2		X			X
Albite	$\text{Na}[\text{AlSi}_3\text{O}_8]$	2.6			X		X
Almandine	$\text{Fe}_3\text{Al}_2[\text{SiO}_4]_3$	4.3		X			X
Amphibole	$(\text{Fe}, \text{Mg}, \text{Ca})_2\text{SiO}_3$	2.9-3.5		X			X
Anatase	$\text{TiO}_2$	3.9			X	X	
Andalusite	$\text{Al}_2\text{SiO}_5$	3.2					X
Andradite	$\text{Ca}_3\text{Fe}_2[\text{SiO}_4]_3$	3.8		X		(2) ←	X
Anhydrite	$\text{CaSO}_4$	3					X
Ankerite	$\text{Ca}_2\text{MgFe}(\text{CO}_3)_4$	2.9-3.1		X			X
Apatite	$\text{Ca}_5(\text{PO}_4)_3(\text{F}, \text{Cl}, \text{OH})$	3.2			X		X
Aragonite	$\text{CaCO}_3$	3			X		X
Arsenopyrite	$\text{FeAsS}$	5.9-6.2		X →	(1)	X	
Asbestos	$\text{Mg}_3[\text{Si}_2\text{O}_5](\text{OH})_4$	2.4-2.5			X		X
Augite	$(\text{Ca}, \text{Na})(\text{Mg}, \text{Fe}, \text{Al}, \text{Ti})(\text{Si}, \text{Al})_2\text{O}_6$	3.2-3.6		X		X →	(1)
Azurite	$\text{Cu}_3(\text{CO}_3)_2(\text{OH})_2$	3.8			X		X
Baddeleyite	$\text{ZrO}_2$	5.5-5.6			X		X
Barite	$\text{BaSO}_4$	4.3-5.0			X		X
Bastnaesite	$(\text{Ce}, \text{La}, \text{F})\text{CO}_3$	4.7-5.0		X			X
Bauxite	$\text{Al}_2\text{O}_3 \cdot n\text{H}_2\text{O}$	2.6			X		X
Beryl	$\text{Be}_3\text{Al}_2[\text{Si}_6\text{O}_{18}]$	2.7-2.8			X		X
Biotite	$\text{K}(\text{Mg}, \text{Fe})_3[\text{Si}_3\text{AlO}_{10}](\text{OH}, \text{F})_2$	2.7-3.1		X			(4)
Bismuth	$\text{Bi}$	9.8			X	X	
Borax	$\text{Na}_2\text{B}_4\text{O}_7 \cdot 10\text{H}_2\text{O}$	1.7			X		X
Bornite	$\text{Cu}_5\text{FeS}_4$	4.9-5.3		(1) ←	X	X	
Brannerite	$(\text{UO}, \text{TiO}, \text{UO}_2)\text{TiO}_3$	4.5-5.4		X		X	
Brookite	$\text{TiO}_2$	4.1			X	X	
Calcite	$\text{CaCO}_3$	2.7			X		X
Carnotite	$\text{K}_2(\text{UO}_2)_2(\text{VO})_4 \cdot 3\text{H}_2\text{O}$	4-5			X	(2) ←	X
Cassiterite	$\text{SnO}_2$	6.4-7.1			X	X	
Celestite	$\text{SrSO}_4$	4			X		X
Cerussite	$\text{PbCO}_3$	6.5			X	(2) ←	X
Chalcocite	$\text{Cu}_2\text{S}$	5.5-5.8			X	X	
Chalcopyrite	$\text{CuFeS}_2$	4.1-4.3		(1) ←	X	X	
Chlorite	$(\text{Mg}, \text{Fe})_3(\text{Si}, \text{Al})_2\text{O}_{10}(\text{OH})_2 \cdot (\text{Mg}, \text{Fe})_3(\text{OH})_6$	2.6-3.3		X			X
Chromite	$(\text{Fe}, \text{Mg})(\text{Cr}, \text{Al})_2\text{O}_4$	4.6		X		X	
Chrysocolla	$(\text{Cu}, \text{Al})_2\text{H}_2\text{Si}_2\text{O}_5(\text{OH})_4 \cdot n\text{H}_2\text{O}$	1.9-2.4			X		X
Cinnabar	$\text{HgS}$	8.0-8.2			X		X
Cobaltite	$(\text{Co}, \text{Fe})\text{AsS}$	6.0-6.3		X		X	
Colemanite	$\text{Ca}_2\text{B}_6\text{O}_{11} \cdot 5\text{H}_2\text{O}$	2.4			X		X
Collophanite	$\text{Ca}_3\text{P}_2\text{O}_8 \cdot \text{H}_2\text{O}$	2.6-2.9			X		(3)
Columbite	$(\text{Fe}, \text{Mn})(\text{Ta}, \text{Nb})_2\text{O}_6$	5.2-8.2		X		X	
Copper	$\text{Cu}$	8.9			X	X	
Corundum	$\text{Al}_2\text{O}_3$	3.9-4.1			X		X
Covellite	$\text{CuS}$	4.6-4.8			X	X	
Cryolite	$\text{Na}_3\text{AlF}_6$	3			X	(2) ←	X
Cuprite	$\text{Cu}_2\text{O}$	5.8-6.2			X		X
Diamond: (natural)	$\text{C}$	3.5			X		X
Diamond: (synthetic)	$\text{C}$	3.5		X			X
Diopside	$\text{CaMgSi}_2\text{O}_6$	3.2-3.5		X →	(1)		X
Dolomite	$\text{CaMg}(\text{CO}_3)_2$	2.8-2.9			X		X
Epidote	$\text{Ca}_2(\text{Al}, \text{Fe})_2\text{Si}_2\text{O}_{12}(\text{OH})$	3.3-3.5		X			X
Euxenite	$(\text{Y}, \text{Ca}, \text{Ce}, \text{U}, \text{Th})(\text{Nb}, \text{Ta}, \text{Ti})_2\text{O}_6$	4.7-5.0		X		X	
Feldspar Group	$(\text{K}, \text{Na}, \text{Ca})(\text{Al}, \text{Si})_3\text{O}_8$	2.5-2.9			X		X
Ferberite	$\text{FeWO}_4$	7.5	(1) ←	X		X	
Flint	$\text{SiO}_2$	2.6			X		X
Fluorite	$\text{CaF}_2$	3.2			X		X
Franklinite	$(\text{Fe}, \text{Mn}, \text{Zn})\text{O} \cdot (\text{Fe}, \text{Mn})_2\text{O}_3$	5.1-5.2	X			X	
Gahnite	$\text{ZnAl}_2\text{O}_4$	4.0-4.6			X		X
Galena	$\text{PbS}$	7.4-7.6			X	X	
Garnet	Complex Ca, Mg, Fe, Mn Silicates	3.1-4.3		X →	(1)	(2) ←	X
Gibbsite	$\text{Al}(\text{OH})_3$	2.3-2.4			X		X
Goethite	$\text{FeO}(\text{OH})$	3.3 - 4.3		X		(2) ←	X
Gold	$\text{Au}$	15.6-19.3			X	X	
Graphite	$\text{C}$	2.1-2.2			X	X	
Grossularite	$\text{Ca}_3\text{Al}_2(\text{SiO}_4)_3$	3.5			X	(2) ←	X
Gypsum	$\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$	2.3			X		X
Halite	$\text{NaCl}$	2.1			X	(2) ←	X
Hematite	$\text{Fe}_2\text{O}_3$	4.9-5.3		X		X	
Hornblende	$\text{Ca}_2(\text{Mg}, \text{Fe}, \text{Al})_3(\text{Al}, \text{Si})_8\text{O}_{22}(\text{OH})_2$	3.1-3.3		X		(2) ←	X
Huebnerite	$\text{MnWO}_4$	6.7-7.5		X →	(1)	X	
Hypersthene	$(\text{Mg}, \text{Fe})\text{SiO}_3$	3.4-3.9		X			X
Ilmenite	$\text{FeTiO}_3$	4.7		X		X	
Ilmenorutile	$\text{Fe}_2(\text{Nb}, \text{Ta})_2 \cdot 4\text{Ti}_{1-x}\text{O}_2$	5.1		X		X	
Ilvaite	$\text{CaFe}^{2+}2\text{Fe}^{3+}\text{Si}_2\text{O}_7\text{O}(\text{OH})$	3.8-4.1		X		X →	(1)
Kaolinite	$\text{Al}_2\text{Si}_2\text{O}_5(\text{OH})_4$	2.2-2.7			X		X
Kyanite	$\text{Al}_2\text{SiO}_5$	3.6-3.7			X		X
Lepidolite	$\text{KLi}_2\text{Al}(\text{Al}, \text{Si})_3\text{O}_{10}(\text{F}, \text{OH})_2$	2.8-2.9			X		X
Leucoxene	$\text{FeTiO}_3 \rightarrow \text{TiO}_2$ (Alteration Product)	3.6-4.3		X →	(1)	X	
Limonite	$\text{FeO}(\text{OH}) \cdot n\text{H}_2\text{O}$	3.6-4.0		X →	(1)	(2) ←	X
Magnesite	$\text{MgCO}_3$	3			X		X
Magnetite	$\text{Fe}_3\text{O}_4$	5.2	X			X	
Malachite	$\text{Cu}_2\text{CO}_3(\text{OH})_2$	3.6-4			X		X
Manganite	$\text{MnO}(\text{OH})$	4.3		X →	(1)	X	

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			F	P	NM	C	NC
Marcasite	FeS <sub>2</sub>	4.8-4.9		(1) ← X		X	
Martite	(See Hematite)	-					
Microline	KAlSi <sub>3</sub> O <sub>8</sub>	2.6			X		X
Microcline	Ca <sub>2</sub> Ta <sub>2</sub> O <sub>7</sub> (See Pyrochlore)	5.5			X		X
Millerite	NiS	5.2-5.6		X		X	
Molybdenite	MoS <sub>2</sub>	4.7-5.0			X	X	
Monazite	(Ce,La,Y,Th)PO <sub>4</sub>	4.9-5.3		X			X
Mullite	Al <sub>4</sub> Si <sub>2</sub> O <sub>13</sub>	3.2			X		X
Muscovite	KAl <sub>2</sub> [AlSi <sub>3</sub> O <sub>10</sub> ](F,OH) <sub>2</sub>	2.8-3.0			X		(4)
Nahcolite	NaHCO <sub>3</sub>	2.2			X		X
Nepheline Syenite	(Na,K)[AlSi <sub>3</sub> O <sub>8</sub> ]	2.6			X		X
Niccolite	NiAs	7.6-7.8		X		X	
Olivine	(Mg,Fe) <sub>2</sub> SiO <sub>4</sub>	3.3-3.5		X			X
Orpiment	As <sub>2</sub> S <sub>3</sub>	3.4-3.5			X	X	
Orthoclase	KAlSi <sub>3</sub> O <sub>8</sub>	2.5-2.6			X		X
Periclase	MgO	3.6-3.9			X		X
Perovskite	CaTiO <sub>3</sub>	4			X		X
Petalite	LiAl(Si <sub>3</sub> O <sub>8</sub> ) <sub>2</sub>	2.4			X		X
Phosphate (pebble)	(See Collaphanite)	-					
Platinum	Pt	14.0-21.5		(1) ← X		X	
Pyrite	FeS <sub>2</sub>	5		(1) ← X		X	
Pyrochlore	(Na,Ca) <sub>2</sub> (Nb,Ti) <sub>2</sub> O <sub>6</sub> (OH,F)	4.2-4.4			X	X	
Pyrolusite	MnO <sub>2</sub>	4.7-5.0		(1) ← X			X
Pyrope	Mg <sub>3</sub> Al <sub>2</sub> (SiO <sub>4</sub> ) <sub>3</sub>	3.5			X	(2) ← X	
Pyroxene	[Ca,Mg,Fe,Al] <sub>2</sub> Si <sub>2</sub> O <sub>6</sub>	3.2-3.6		X → (1)		(2) ← X	
Pyrrhotite	Fe <sub>1-x</sub> S <sub>x</sub>	4.6-4.7	X			X	
Quartz	SiO <sub>2</sub>	2.7			X		(3)
Realgar	AsS	3.6			X	X	
Rhodochrosite	MnCO <sub>3</sub>	3.4-3.7			X	(2) ← X	
Rhodonite	MnSiO <sub>3</sub>	3.6-3.7			X	(2) ← X	
Rutile	TiO <sub>2</sub>	4.2-4.3			X	(2)	
Samarskite	(Y,Ce,U,Fe) <sub>3</sub> (Nb,Ta,Ti) <sub>3</sub> O <sub>16</sub>	4.3 - 5.87	(1) ← X			X	
Scheelite	CaWO <sub>4</sub>	5.9-6.1			X		X
Serpentine	Mg <sub>3</sub> [Si <sub>4</sub> O <sub>10</sub> ](OH) <sub>8</sub>	2.2-2.7		X			X
Siderite	FeCO <sub>3</sub>	3.8		X		(2) ← X	
Sillimanite	Al <sub>2</sub> O[SiO <sub>4</sub> ]	3.2			X		X
Silver	Ag	10.1-11.1			X	X	
Smithsonite	ZnCO <sub>3</sub>	4.3-4.5			X		X
Sodalite	Na <sub>8</sub> [Al <sub>6</sub> Si <sub>6</sub> O <sub>24</sub> ]Cl <sub>2</sub>	2.1-2.3			X		X
Spessartine	Mn <sub>2</sub> Al <sub>2</sub> (SiO <sub>4</sub> ) <sub>3</sub>	4.2			X		X
Sphalerite	(Zn,Fe)S	3.9-4.1		X → (1)		X → (1)	
Sphene (Titanite)	CaTiSiO <sub>5</sub>	3.4-3.6			X	(2) ← X	
Spinel	MgAl <sub>2</sub> O <sub>4</sub>	3.6		(1) ← X		X	(1)
Spodumene	LiAl(SiO <sub>3</sub> ) <sub>2</sub>	3.1-3.2			X		X
Stannite	Cu <sub>2</sub> FeSnS <sub>4</sub>	4.3-4.5			X	X	
Staurolite	Fe <sup>2+</sup> Al <sub>4</sub> [Si <sub>4</sub> O <sub>11</sub> ] <sub>2</sub> O <sub>2</sub> (OH) <sub>2</sub>	3.6-3.8		X		(2) ← X	
Stibnite (Antimonite)	Sb <sub>2</sub> S <sub>3</sub>	4.6			X	X	
Struverite	(Ta <sub>2</sub> O <sub>5</sub> ,Nb <sub>2</sub> O <sub>5</sub> ) <sub>n</sub> TiO <sub>2</sub>	5.1		X		X	
Sulphur	S	2.1			X		X
Sylvite	KCl	2			X		X
Talc	Mg <sub>3</sub> Si <sub>4</sub> O <sub>10</sub> (OH) <sub>2</sub>	2.5-2.8			X		X
Tantalite	(Fe,Mn)(Ta,Nb) <sub>2</sub> O <sub>6</sub>	7.3-7.8		X		X	
Tapiolite	Fe,Mn (Ta,Nb) <sub>2</sub> O <sub>6</sub>	4.6-5.2		X		X	
Tetrahedrite	(Cu,Fe,Zn) <sub>12</sub> Sb <sub>4</sub> S <sub>13</sub>	5		X		X	
Thorianite	ThO <sub>2</sub>	9.7			X		X
Thorite	ThSiO <sub>4</sub>	4.5-5.4			X		X
Topaz	Al <sub>2</sub> SiO <sub>4</sub> (F,OH) <sub>2</sub>	3.4-3.6			X		X
Tourmaline	(Na,Ca)[Mg,Fe <sup>2+</sup> ,Fe <sup>3+</sup> ,Al,Li] <sub>3</sub> Al <sub>3</sub> (BO <sub>3</sub> ) <sub>3</sub> Si <sub>6</sub> O <sub>18</sub> (OH) <sub>4</sub>	2.9-3.2		X → (1)		(1,2) ← X	
Uraninite	UO <sub>2</sub>	9-9.7		X			X
Vermiculite	(K,Fe)Mg <sub>3</sub> [Al <sub>2</sub> Si <sub>4</sub> O <sub>10</sub> ](OH) <sub>2</sub> •nH <sub>2</sub> O	2.4-2.7			X		X
Wolframite	(Fe,Mn)WO <sub>4</sub>	6.7-7.5		X		X	
Wollastonite	CaSiO <sub>3</sub>	2.8-2.9			X		X
Wulfenite	PbMoO <sub>4</sub>	6.5-7.0			X	X	
Xenotime	YPO <sub>4</sub>	4.4-5.1		X			X
Zeolite	Hydrous Alumino-Silicate usually of Ca and Na	2.0-2.5			X		X
Zincite	ZnO	5.7			X	(1) ← X	
Zircon	ZrSiO <sub>4</sub>	4.7			X		X

#### Explanation of Table:

The magnetic and electrostatic separation characteristics listed in this table are based on separation tests made in Outotec's laboratory.

Magnetic Response is classified as (a) ferromagnetic: attraction to a conventional permanent magnetic separator at a metric flux density of less than 0.2 Tesla (2,000 gauss), (b) paramagnetic: attraction to the magnetic zone of a high intensity magnetic separator operating at greater than 0.2 to 2.0 Tesla (2,000-20,000 gauss), and (c) non-magnetic or diamagnetic: not-affected or repelled respectively from the magnetic force described in (b).

Electrostatic response is classified as conductive or non-conductive based on the surface resistivity of a mineral. Generally, minerals with a surface resistivity of less than 1011 ohms will report as "conductors" using standardized settings on an electrostatic separator. Minerals with a surface resistivity above 1012 ohms will report as "non-conductors" under similar equipment settings.

**Notes:** (1) Variations in response to magnetic and electrostatic separation will occur based on the actual composition of the mineral from varying locations and the presence of inclusions, (2) The response of some minerals can be varied by treatment at elevated temperature, i.e. rutile becomes more conductive at temperatures greater than 200 C, (3) Triboelectric (contact) charging can be used to selectively charge certain minerals; and they, in turn, can be attracted to an electrode of the opposite polarity, i.e. quartz from colophonite (pebble phosphate), (4) The separation of particles with a pronounced shape factor (mica) can also be accomplished by the use of a non-uniform electric field (dielectrophoresis).

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email:psdsales@outotec.com  
www.outotec.com